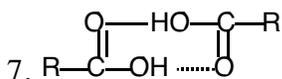
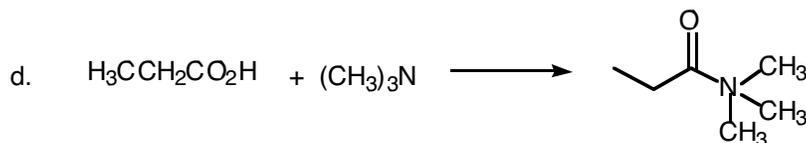
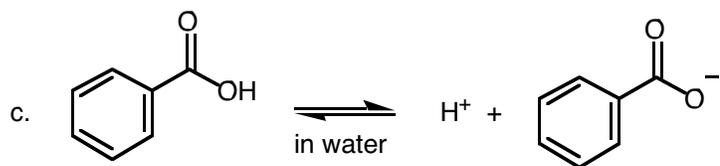
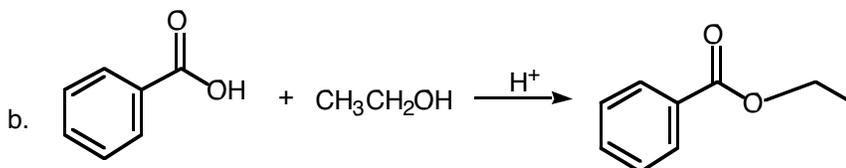
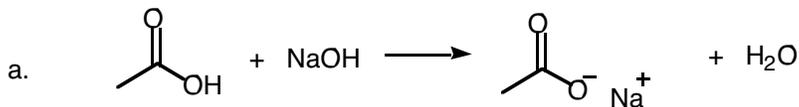


Answer Key for Drill sheet 3

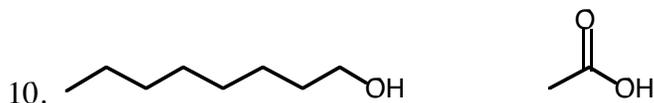
1. C. 2. C 3. A 4. B 5. A

6.

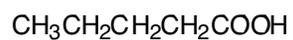


8. Carboxylic acid > alcohol > ester > alkane

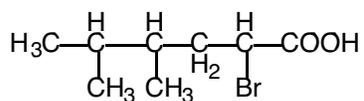
9. hydrogen bonding between molecules. The greater the hydrogen bonding between molecules, the more energy it requires to vaporize the liquid and therefore the higher is the boiling point.



11.

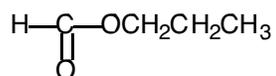
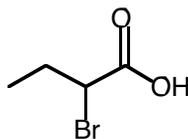


pentanoic acid

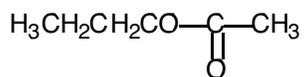


2-bromo-4,5-dimethylhexanoic acid

2-bromobutanoic acid

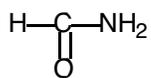
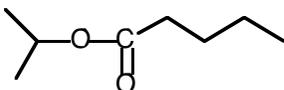


propyl formate

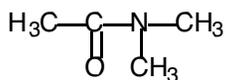


propyl acetate

isopropyl pentanoate



formamide



*N,N*-dimethyl acetamide

*N*-methylpropanamide

